

## TRADE (TECHNICAL) TEST QUESTIONS – QAPCO, QATAR

1. In a desorption column, which tray is the sensitive tray?

**Answer:** Top tray where temperature is controlled.

2. Why is CBD carried out in boilers?

**Answer:** For better steam quality.

3. In surging, what is the main cause?

**Answer:** Suction low flow.

4. Which parameter is checked in wastewater effluent?

**Answer:** Chemical Oxygen Demand (COD).

5. Why is demineralized water pH kept slightly basic?

**Answer:** To avoid corrosion.

6. Polymerization process is:

**Answer:** Exothermic.

7. 6 g of carbon is equal to:

**Answer:** 0.5 mole.

8. In case of fire, first action is:

**Answer:** Inform CCR.

9. In case of sulfur smell, first action is:

**Answer:** Use escape mask and inform CCR.

10. Forklift entry in operating area requires:

**Answer:** Valid work permit.

11. Why is saturated steam used in reboilers?

**Answer:** To reduce heat duty.

12. Boiling point is:

**Answer:** Temperature at which vapor pressure equals system pressure.

13. What is flooding?

**Answer:** High accumulated level in the column.

14. Who authorizes scaffolding?

**Answer:** Safety section.

15. What is cavitation?

**Answer:** When NPSH available < NPSH required.

16. How is flow controlled in a centrifugal pump?

**Answer:** By throttling the discharge valve.

17. Which action is wrong to control flow in a PD compressor?

**Answer:** Throttling the discharge valve.

18. What is a volumetric pump?

**Answer:** Positive displacement (PD) pump.

19. Why are polymers solid at ambient temperature?

**Answer:** Due to higher molecular weights.

20. Which statement is not a quality policy?

**Answer:** Maximum production.

21. Which is not a quality policy?

**Answer:** Traceability of product is optional.

22. What shows chemical properties?

**Answer:** MSDS (Material Safety Data Sheet).

23. Which is saturated steam?  
**Answer:** Steam at saturated pressure.

24. Why is saturated steam used?  
**Answer:** Due to latent heat.

25. How is demineralized water conductivity checked?  
**Answer:** Because it is a good conductor after ion-exchange process.

26. How is viscosity checked?  
**Answer:** By allowing liquid through a calibrated diameter orifice.

27. Why is fast purging done in boilers?  
**Answer:** To remove sludge and dirt.

28. Which are greenhouse gases?  
**Answer:** CO<sub>2</sub> and methane.

29. Why are hosepipe couplings kept flexible?  
**Answer:** To facilitate operator handling.

30. What is the color of mandatory signs?  
**Answer:** Blue.

31. What are disadvantages of sulfur dust?  
**Answer:** Irritation and toxicity.

32. What does latent heat do?  
**Answer:** Causes phase change while temperature remains constant.

33. Is the ethylene process fast or slow?  
**Answer:** Fast.

34. What is the meaning of "correction"?  
**Answer:** Rework.

35. Demin water pH:  
**Answer:** 8 (basic).

36. Why is steam turbine heating carried out?  
**Answer:** To avoid thermal shock.

37. Why is O<sub>2</sub> kept in excess in a furnace?  
**Answer:** For complete combustion.

38. In a refrigeration compressor, if suction pressure goes high, what happens?  
**Answer:** Cooling reduces and process temperature rises.

39. In solid handling, what is used for dilute phase conveying?  
**Answer:** Air at 3 bar pressure.

40. Why are double block and bleed valves used?  
**Answer:** To isolate the system.

41. How are N<sub>2</sub> leakages detected?  
**Answer:** None of these (no odor, color, or smell).

42. Why are additives used in polymers?  
**Answer:** For customer satisfaction.

43. Which affects polyolefins more: oxygen, water, or acetylene?  
**Answer:** Acetylene.

44. If O<sub>2</sub> is 21% in a container under atmospheric pressure, what is O<sub>2</sub> concentration if the same container is pressurized with 6 kg/cm<sup>2</sup> nitrogen?  
**Answer:** 3.5%

45. Some chemical equations to write on answer sheet:

- $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- $7\text{O}_2 + 2\text{C}_2\text{H}_6 \rightarrow 4\text{CO}_2 + 6\text{H}_2\text{O}$
- $2\text{NaOH} + \text{CO}_2 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O}$
- $2\text{NaOH} + \text{H}_2\text{S} \rightarrow \text{Na}_2\text{S} + 2\text{H}_2\text{O}$